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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/538,079	06/09/2005	Koji Matsumoto	0020-5381PUS1	7194

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EXAMINER
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MILLER, MICHAEL G

ART UNIT	PAPER NUMBER
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1792

NOTIFICATION DATE	DELIVERY MODE
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03/19/2008

ELECTRONIC

**Please find below and/or attached an Office communication concerning this application or proceeding.**

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

mailroom@bskb.com

<b>Office Action Summary</b>	<b>Application No.</b> 10/538,079	<b>Applicant(s)</b> MATSUMOTO ET AL.	
	<b>Examiner</b> MICHAEL G. MILLER	<b>Art Unit</b> 1792	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

### Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

### Status

- 1) ☒ Responsive to communication(s) filed on 09 June 2005.
- 2a) ☐ This action is **FINAL**.                      2b) ☒ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

### Disposition of Claims

- 4) ☒ Claim(s) 1-10 is/are pending in the application.
- 4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.
- 5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.
- 6) ☒ Claim(s) 1-10 is/are rejected.
- 7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.
- 8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

### Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on \_\_\_\_\_ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

### Priority under 35 U.S.C. § 119

- 12) ☒ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☒ All    b) ☐ Some \*    c) ☐ None of:
1. ☒ Certified copies of the priority documents have been received.
2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

### Attachment(s)

- |  |   |
|--|---|
| 1) <input checked="" type="checkbox"/> Notice of References Cited (PTO-892)            | 4) <input type="checkbox"/> Interview Summary (PTO-413)           |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948)   | Paper No(s)/Mail Date. _____                                      |
| 3) <input checked="" type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08) | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| Paper No(s)/Mail Date <u>9 Jun 2005, 23 Oct 2006</u>                                   | 6) <input type="checkbox"/> Other: _____                          |

## DETAILED ACTION

### ***Claim Rejections - 35 USC § 103***

- 1) The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.
- 2) The factual inquiries set forth in *Graham v. John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
  - (1) Determining the scope and contents of the prior art.
  - (2) Ascertaining the differences between the prior art and the claims at issue.
  - (3) Resolving the level of ordinary skill in the pertinent art.
  - (4) Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 3) This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

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- 4) Claims 1-3, 5, and 7-10 are rejected under 35 U.S.C. 103(a) as being unpatentable over Isozaki (U.S. Patent 6,337,369, hereinafter '369) in view of Starzewski (U.S. Patent 5,670,092, hereinafter '092).
- 5) With regard to Claim 1, '369 teaches a method for producing a polarizing film comprising:
  - a) The step of dipping a polyvinyl alcohol film in/on which iodine is adsorbed and oriented in an aqueous solution containing boric acid ('369 PG 0038).
  - b) '369 is silent as to the limitation wherein contact between the aqueous solution and oxygen is suppressed.
  - c) '369 teaches that for PVA dry-stretching, an oxygen-poor atmosphere is desirable to prevent discoloration of the PVA (Column 4 Lines 1-12) and that a heat treatment may be conducted after the fixing step (Column 4 Lines 59 - 67).
  - d) '092 teaches that a post-fixing heat treatment improves the polarization properties of the PVA film when performed in the absence of oxygen. (Column 2 Lines 4-7, Column 3 Line 66 – Column 4 Line 31).
  - e) Therefore, it would have been obvious to a person having ordinary skill in the art at the time the invention was made to have combined the method of '369 with the post-treatment step of '092 because '369 wants to create a polarizing film and '092 teaches a way to improve the optical properties of a polarizing film.
  - f) '369/'092 discloses the claimed invention except for wanting to suppress contact between the aqueous solution and oxygen. It would have been obvious to one having ordinary skill in the art at the time the invention was made to perform this

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suppression since it was known in the art that oxygen produces deleterious effects in the processing steps immediately surrounding it (discoloration in the stretching step, impaired polarization in the heat treatment step).

6) With regard to Claim 2, '369/'092 teaches the method according to claim 1, except for the following limitation:

- a) The contact between said aqueous solution containing boric acid and oxygen is suppressed by adjusting an oxygen concentration in an atmosphere which is in contact with said aqueous solution to 5% by volume or less.
- b) Both '369 and '092 teach using nitrogen atmospheres to exclude oxygen during the dry stretching and heat treatment respectively. There would be a reasonable expectation of success of excluding oxygen from contact with the aqueous solution if a nitrogen atmosphere were kept over the aqueous solution. The degree of exclusion is a result-effective variable and as such is held to be routine experimentation.

7) Claim 3 is duly rejected on the same grounds as Claim 2, as Claim 3 calls for the method of either Claim 1 or Claim 2 wherein the exclusion is performed using an inert gas (nitrogen taught above).

8) Claim 5 is rejected on the same grounds as Claim 3, as it requires the method of Claim 3 wherein the inactive gas is nitrogen.

9) With regard to Claim 7, '369/'092 teaches the method according to claim 1, wherein:

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- a) A weight ratio of water : boric acid : potassium iodide in said aqueous solution containing boric acid is 100 : (2 – 15) : (2 - 20) ('369 Column 6 Lines 38 – 51 teaches 100 : 2 : 4).

10)With regard to Claim 8, '369/'092 teaches the method according to claim 1, wherein:

- a) A dipping time is from 90 to 1,200 seconds ('369 Column 6 Lines 38-51, 240s).
- b) '369 teaches a temperature of 30°C for the aqueous solution which is below the claimed temperature of 50 to 85°C.
- c) '369/'092 discloses the claimed invention except for the specific temperature of the fixing bath. It would have been obvious to a person having ordinary skill in the art at the time the invention was made to adjust the temperature of the bath, since it has been held that where the general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re Aller*, 220 F.2d 454, 105 USPQ 223 (CCPA 1955).

11)With regard to Claim 9, '369/'092 teaches the method according to claim 1, wherein

said polyvinyl alcohol film has a polymerization degree of 1,500 to 5,000 ('369 Column 6 Lines 38-51, 4000).

12)With regard to Claim 10, '369/'092 teaches the method according to claim 1, wherein

said polyvinyl alcohol film in/on which iodine is adsorbed and oriented is a film produced by dipping an unstretched polyvinyl alcohol film in a solution containing iodine and potassium iodide and then uniaxially stretching it ('369 Column 6 Lines 38-51).

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13) Claim 4 is rejected under 35 U.S.C. 103(a) as being unpatentable over '369/'092 as applied to claim 3 above, and further in view of DesMarais et al (U.S. Patent 6,362,244, hereinafter '244).

14) With regard to Claim 4, '369/'092 teaches the method according to claim 3 except for the following limitation:

- a) Wherein the dipping of said polyvinyl alcohol film in said aqueous solution containing boric acid is carried out while bubbling said inactive gas in said aqueous solution.
- b) As shown above, '369/'092 has motivations to remove oxygen from the processing system. It known in nature that aqueous solutions can carry dissolved gases, including oxygen.
- c) '244 teaches that deoxygenation of a liquid is known and that is usually performed by sparging a liquid with nitrogen or argon (Column 5 Lines 1-5).
- d) Therefore, it would have been obvious to a person having ordinary skill in the art at the time the invention was made because '369/'092 wants to make a polarizing film in a low-oxygen environment and '244 teaches that one way to remove oxygen from a liquid system is to bubble an inert gas through the system.

15) Claim 6 is rejected under 35 U.S.C. 103(a) as being unpatentable over '369/'092 as applied to claim 1 above, and further in view of Dempo (U.S. Patent 5,512,178, hereinafter '178).

16) With regard to Claim 6, '700 teaches the method according to claim 1, except for the following limitation:

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- a) Wherein said polyvinyl alcohol film is dipped in said aqueous solution containing boric acid while said aqueous solution is treated with activated carbon continuously or intermittently.
- b) As shown above, '369/'092 has motivations to remove oxygen from the processing system. It known in nature that aqueous solutions can carry dissolved gases, including oxygen, as well as other oxidizing materials.
- c) '178 teaches that an activated carbon filter can be used to remove oxidizing substances from an aqueous solution (Column 3 Lines 34 – 48). Oxygen is the prototypical oxidizing substance in nature.
- d) Therefore, it would have been obvious to a person having ordinary skill in the art at the time the invention was made because '369/'092 wants to make a polarizing film in a low-oxygen environment and '178 teaches that one way to remove oxygen from a liquid system is to bubble pass the liquid through an activated carbon filter.

### ***Conclusion***

Any inquiry concerning this communication or earlier communications from the examiner should be directed to MICHAEL G. MILLER whose telephone number is (571)270-1861. The examiner can normally be reached on M-F 7-4.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Tim Meeks can be reached on (571) 272-1423. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.



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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

MGM

/Michael G. Miller/  
Examiner, Art Unit 1792

/Timothy H Meeks/  
Supervisory Patent Examiner, Art Unit 1792